APPENDIX E

NICEATM Summary of The Multicenter Evaluation of *In Vitro* Cytotoxicity (MEIC)

This document was provided in the Background Materials and Supplemental Information Notebook for the International Workshop on *In Vitro* Methods for Assessing Acute Systemic Toxicity [Section I, TAB 6].

The following ATLA (Alternatives To Laboratory Animals) excerpts are reprinted with permission from Professor Michael Balls, editor of ATLA.

- Clemedson et al., 1998. MEIC Evaluation of Acute Systemic Toxicity, Part IV. ATLA 26: 131-183. **[Table 1]**
- Ekwall et al., 1998. MEIC Evaluation of Acute Systemic Toxicity, Part V. ATLA 26: 571-616. [Tables II, III, IV, V, VI, IX]
- Ekwall et al., 2000. MEIC Evaluation of Acute Systemic Toxicity, Part VIII, ATLA 28 Suppl 1, 201-234. [Figures 1 and 10]
- Ekwall et al., 1999. EDIT: A new international multicentre programme to develop and evaluate batteries of in vitro tests for acute chronic systemic toxicity. ATLA 27: 339-349. [Table 1 and Figure 1]

The following table was reproduced with permission from Dr. Gary Hook (NIEHS).

• Wallum, E. 1998. Acute Oral Toxicity. EHP 106: 497-503. [reproduction of Table 1]

The Multicenter Evaluation of *In Vitro* Cytotoxicity (MEIC)

Summary

September 2000

National Toxicology Program (NTP) Interagency Center for the Evaluation of Alternative Toxicological Methods (NICEATM)

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1.0 Introduction

The Multicenter Evaluation of In Vitro Cytotoxicity (MEIC) program was organized by the Scandinavian Society for Cell Toxicology in 1989. MEIC was started with two goals. The first was to investigate the relevance of results from in vitro tests for predicting the acute toxic action of chemicals in humans. The second was to establish batteries of existing *in vitro* toxicity tests as replacements for acute toxicity tests on animals (LD50). Achievement of the second goal, the practical and ethical one, was considered to be entirely dependent on a successful outcome of the first, scientific goal. At the same time, it was recognized that a demonstrated high relevance of in vitro toxicity tests for human acute toxicity did not mean that all problems of replacement of animal tests would be solved. MEIC was a voluntary effort involving 96 international laboratories that evaluated the relevance and reliability of in vitro cytotoxicity tests originally developed as alternatives to or supplements for animal tests for acute systemic toxicity, chronic systemic toxicity, organ toxicity, skin irritancy, or other forms of general toxicity. In establishing the framework for this program, a minimum of methodological directives was provided in order to maximize protocol diversity among the participating laboratories. The collection of test method data was completed in 1996. multiple publications originating from these studies are provided in chronological order in All in vitro toxicity test results Section 12. collected during MEIC are available on the Cytotoxicology Laboratory, Uppsala (CTLU) website (www.ctlu.se) as a searchable database.

2.0 Test Chemicals

Fifty reference chemicals were selected for testing (Appendix 1). Selection was based on the availability of reasonably accurate human data on acute toxicity. Due to the anticipated five-year duration of MEIC, it was recognized that multiple samples (lots) of each chemical would be needed. However, it was decided that the chemicals would not be provided by a central supplier, but rather that each laboratory would purchase each chemical at the highest purity obtainable with the

proviso that storage duration would be kept to a minimum. The decision to not have a central supplier was based on the rationale that most reference chemicals are drugs, which presents fewer impurity problems. It is also based on the recognition that the results would be evaluated against human poisonings, which involve chemicals of different origin and purity.

3.0 In Vitro Test Assays

By the end of the project in 1996, 39 laboratories had tested the first 30 reference chemicals in 82 *in vitro* assays, while the last 20 chemicals were tested in 67 *in vitro* assays (**Appendix 2**). Slight variants of four of the assays were also used to test some chemicals. The primary 82 assays included:

- Twenty human cell line assays utilizing Chang liver, HeLa, Hep 2, Hep G2, HFL1, HL-60, McCoy, NB-1, SQ-5, and WI-1003 cells:
- Seven human primary culture assays utilizing hepatocytes, keratinocytes, and polymorphonuclear leukocytes;
- Nineteen animal cell line assays utilizing 3T3, 3T3-L1, Balb 3T3, BP8, ELD, Hepa-1c1c7, HTC, L2, LLC-PK1, LS-292, MDBK, PC12h, and V79 cells;
- Eighteen animal primary culture assays utilizing bovine spermatozoa, chicken neurons, mouse erythrocytes, rat hepatocytes, and rat muscle cells; and
- Eighteen ecotoxicological tests utilizing bacteria (Bacillus subtilis, Escherichia coli B, Photobacterium phosphoreum, Vibrio fisheri), rotifer (Brachionus calyciflorus), crustacea (Artemia salina, Daphnia magna, Streptocephalus proscideus), plant (Alium cepa root, tobacco plant pollen tubes), and fish (trout hepatocytes, trout R1 fibroblast-like cells).

4.0 Assay Endpoints

The analyses conducted by the MEIC management team were based on *in vitro* toxicity data presented as IC50 values (i.e., the dose

estimated to reduce the endpoint in question by 50%) (**Appendix 2**).

These values were generated by the participating laboratories and were not independently verified; original data were not presented in the MEIC publications. Thirty-eight of these assays were based on viability, 29 on growth, and the remaining assays involved more specific endpoints, such as locomotion, contractility, bioluminescence. motility. velocity. immobilization. The endpoints assessed were based on exposure durations ranging from five minutes to six weeks, and included:

- Cell viability as measured by the metabolism of 3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2H tetrazolium bromide (MTT), neutral red uptake (NRU), lactate dehydrogenase (LDH) release, cell morphology, adenosine triphosphate (ATP) content or leakage, trypan blue exclusion, viable cell count, tritiated-proline uptake, 86Rb leakage, creatine kinase activity, and glucose consumption;
- Cell growth as measured by protein content, macromolecule content, cell number, pH change, and optical density;
- Colony formation as measured by plating efficiency;
- An organotypic cellular endpoint (i.e., contractility of rat skeletel muscle cells);
- Motility and velocity for bovine sperm;
- Bioluminescence; and
- Mortality in lower eukaryotic organisms.

5.0 Comparative Data

The types of comparative data used to evaluate the predictive accuracy of the *in vitro* IC50 toxicity data for human acute toxicity included:

Oral rat and mouse LD50 values obtained from Registry of Toxic Effects of Chemical Substances (RTECS) (Appendix 3, which contains rat and mouse LD50 data and average human lethal dose data for the 50 MEIC chemicals, ranked in three consecutive tables according to potency for rat, then

- mouse, and finally human. It also contains an U.S. Environmental Protection Agency (EPA) classification scheme for the acute toxicity of chemicals in humans.);
- Acute oral lethal doses in humans obtained from nine reference handbooks (Appendix 4);
- Clinically measured acute lethal serum concentrations in humans obtained from ten reference handbooks (Appendix 5);
- Acute lethal blood concentrations in humans measured post-mortem obtained from one forensic handbook and six forensic tabulations (**Appendix 6**);
- Human pharmacokinetics following single doses, including absorption, peak time, distribution/elimination curves, plasma half-life, distribution volume, distribution to organs (notably brain), and blood protein binding (**Appendix 7**);
- Peaks from curves of an ~50% lethal blood/serum concentration over time after ingestion (LC50 curves derived from human acute poisoning case reports) (Appendix 8);
- Qualitative human acute toxicity data, including lethal symptoms, main causes of death, average time to death, target organs, presence of histopathological injury in target organs, presence of toxic metabolites, and known or hypothetical mechanisms for the lethal injury (Appendix 9).

Early in the MEIC project, the in vitro cytotoxicity results were compared with average lethal blood concentrations (LCs) from acute human poisoning. However, these LCs were of limited value because they were averages of data with a wide variation due to different time between exposure and sampling (clinical) or death (forensic medicine). Therefore, a project was started to collect published and unpublished (from poison information centers and medico-legal institutes) case reports from human poisonings for the 50 MEIC reference chemicals that had lethal or sublethal blood concentrations with known time between ingestion and sampling/death. The aim was to compile enough case reports to be able to construct time-related lethal concentration

curves to be compared with the IC50 values for different incubation times in vitro. The results from the project were presented and analyzed in a series of 50 MEIC monographs. All monographs with sufficient case reports contain five tables presenting blood concentrations and two figures presenting LC curves. Three tables present (i) clinically measured, time-related sublethal blood concentrations, (ii) clinically measured, timerelated lethal blood concentrations, and (iii) postmortem, time-related blood concentrations. In these tables, blood concentration and the time interval between exposure and sampling for these concentrations are listed, as well as other important information on the cases. One table contains case reports with blood concentrations without a known time after ingestion and one table presents average blood concentrations calculated from the values presented in the other tables. The two figures presented in each of the monographs are scatter plots of sublethal and lethal blood concentrations. Based on these plots, concentration curves over time were drawn for the highest no lethal concentrations (NLC100); the lowest lethal concentrations (LC0); and the median curve between NLC100 and LC0, which is called the approximate LC50 even though it is not equivalent to a 50% mortality.

6.0 Statistical Analyses

The statistical analyses conducted by the MEIC management team involved:

- Principal components analysis (PCA);
- Analysis of Variance (ANOVA) and pairwise comparison of means using Tukey's method;
- Linear regression and ANOVA linear contrast analysis; and
- Multivariable partial least square (PLS) modeling with latent variables.

7.0 Results (based on IC50 response)

The MEIC management team, based on their analyses of the *in vitro* IC50 data, obtained the following results:

• The 1st PCA component described 80% of the variance of all the cytotoxicity data.

- Tukey's ANOVA indicated a similar sensitivity (~80%) for the assays.
- The toxicity of many chemicals increased with exposure time, making it necessary to perform a test at several exposure times to fully characterize the cytotoxicity.
- In general, human cytotoxicity was predicted well by animal cytotoxicity.
- Prediction of human cytotoxicity by ecotoxicological tests was only fairly good.
- One organotypic endpoint (muscle cell contractility) gave different results to those obtained with viability/growth assays.
- Sixteen comparisons of similar test systems involving different cell types and exposure times revealed similar toxicities, regardless of cell type.
- Nine of ten comparisons of test systems with identical cell types and exposure times revealed similar toxicities, regardless of the viability or growth endpoint measurement used.
- Nine comparisons of similar test systems employing different primary cultures and cell lines indicated that they shared similar toxicities.
- A high correlation between an intracellular protein denaturation test and average human cell line toxicity test suggested that denaturation may be a frequently occurring mechanism in basal cytotoxicity.

The following results were based on comparisons between *in vitro* data and *in vivo* data:

- Simple human cell tests were shown to be relevant for human acute lethal action for as many as 43 of the 50 MEIC reference chemicals (86%). The exceptions were atropine, digoxin, malathion, nicotine, cyanide, paracetamol, and paraquat -- all specific receptor-mediated toxicants.
- A battery of three of these human cell line tests (nos. 1, 9, 5/16) was found to be highly predictive ($R^2 = 0.77$) of the peak human lethal blood concentrations (LC50) of chemicals. The prediction increased markedly ($R^2 = 0.83$) when a simple

algorithm based on the knowledge of passage across the blood-brain barrier was used to adapt in vitro to in vivo concentrations (**Appendix 7**). The battery involved four endpoints and two exposure times (protein content/24 hours; ATP content/24 hours; inhibition of elongation of cells/24 hours; pH change/7 days). Prediction was better than the prediction of human lethal doses by rat and mouse LD50-values ($R^2 = 0.65$). The correlation between calculated oral LD50 doses in rats and mice and acute lethal dose in humans is presented graphically in **Appendix 10**, while the correlation between IC50 values and peak lethal blood concentrations in humans is presented graphically in **Appendix 11**.

- In the in vitro -- in vivo MEIC evaluation of chemicals that do easily not cross the blood-brain barrier, the 24 hour cytotoxic concentrations for rapidly acting chemicals correlated well with the human lethal peak blood concentrations, while the corresponding cytotoxicity for the slow-acting chemicals did not correlate as well with the peak concentrations. The prediction of human toxicity by the tests of slow-acting chemicals was much improved when 48-hour cytotoxic concentrations were compared with 48hour human lethal blood concentrations. Thus, an in vitro test providing a discrimination between a rapid and a slow cytotoxic action would increase the predictive power of a cell test battery on acute toxicity.
- The findings from both the *in vitro-in vitro* comparisons and the *in vitro-in vivo* comparisons strongly supported the basal cytotoxicity concept.

8.0 MEIC Conclusions and Recommendations

Based on the analyses conducted, the MEIC management team made the following conclusions:

• The MEIC 1, 9, 5/16 test battery can be used directly as a surrogate for a LD50

test. However, since the battery predicts lethal blood concentrations, not lethal dosages, it is not a direct counterpart of the animal LD50 test. Thus, the 1, 9, 5/16 battery must be supplemented with data on gut absorption as well as the distribution volumes (Vd) of chemicals. Vd essentially depends on whether chemicals penetrate cells or not, and the degree of accumulation in the cell for chemicals that enter cells. Binding to proteins, lipids, bone and intracellular matrix will also influence Vd. Probably, a simple test of accumulation in cells over time would provide adequate Vd data. There is sufficient *knowledge of kinetics and Vd to enable an evaluation of results from such an assay for most of the 50 MEIC chemicals.

- An ongoing evaluation is being conducted to address the issue of predicting human oral lethal doses rather than human lethal blood concentrations. One MEIC manuscript in preparation will focus on the importance of the kinetic determinants of target organs for basal cytotoxicity. A second MEIC manuscript will describe how human lethal doses may be predicted by cellular tests on basal cytotoxicity (the 1, 9, 5/16 battery) and kinetic data.
- If human lethal doses are shown to be well predicted by the 1, 9, 5/16 battery, when combined with absorption and distribution data, a new but simple *in vitro* test to predict distribution volumes must be developed. An effective *in vitro* test on absorption is stated to already exist. Development of new *in vitro* methods is not addressed by MEIC, which only evaluated existing methods.
- In MEIC, only two of the 50 reference chemicals (ethylene glycol and methanol) were biotransformed to more toxic metabolites, contributing to the acute lethal action. The occurrence of toxic metabolites for the two chemicals did not affect the prediction of human lethal peak concentrations by human cell line inhibitory concentrations, but seemed to interfere with the correlation between *in vitro* delayed effects and the prediction of

later lethal effects of the chemicals. These results confirm the proposed usefulness of an *in vitro* test that could measure the formation and release of a toxic metabolite by metabolically competent cells within the time frame of acute toxicity. One design of such a test would be to use human hepatocytes in cocultures with a target cell line. Since so few metabolically active chemicals were tested in MEIC, future studies will need to include additional metabolically activated chemicals.

9.0 Evaluation-Guided Development of *In Vitro* Tests (EDIT)

In recognition that additional *in vitro* tests were needed to enhance the accuracy of the proposed *in vitro* battery for predicting human acute toxicity, a second voluntary multicenter program was initiated by the CTLU. The CTLU has designed a blueprint for an extended battery and has invited all interested laboratories to develop the "missing" tests of this battery within the

framework of the EDIT program (Appendix 12 and 13). The EDIT research program is published on the Internet (www.ctlu.se). The aim of EDIT is to provide a full replacement of the animal acute toxicity tests. The most urgently needed developments are assays on the accumulation of chemicals in cells (test of Vd), passage across the intestinal and blood-brain barriers, biotransformation to more toxic metabolites. CTLU will provide interested laboratories with human reference data and will evaluate results as single components of complex models. Internet version of the general EDIT research program contains additional, regularly updated information on the project. Purported advantages of the project are as follows. First, the evaluationguided test development in EDIT is rational since tests are designed according to obvious needs and as elementary tests of single events integrated into whole models, which is the potential strength of the in vitro toxicity testing strategy. Second, the direct testing of MEIC chemicals in newly developed in vitro assays will lead to a rapid evaluation of the potential value of each assay.

10.0 Recommended Integration of MEIC/EDIT into the EPA High Production Volume (HPV) Program

Dr. Ekwall, the principle scientist for the MEIC program, has provided several suggestions for using MEIC results and the forthcoming EDIT results to reduce animal testing in the HPV program. These suggestions include the following:

- 1. Formal validation by ECVAM/ICCVAM of the existing 3 test MEIC battery. If considered validated, use of the battery to test every chemical in the HPV program would provide inexpensive and useful supplementary data.
- 2. Evaluate some of the HPV chemicals in a battery of *in vitro* toxicity and toxicokinetic tests on acute toxicity (EDIT and similar models) as follows:
 - Engage poison information experts to select a set of HPV chemicals with sound human acute toxicity data, including time-related lethal blood concentrations.
 - Give priority to standard testing of the same chemicals in the HPV program.
 - Testing of the same chemicals in the newly developed *in vitro* systems (EDIT, etc.), including modeling of acute toxicity by the new assays.
 - Comparison of HPV standard animal data and the *in vitro* data with the human data for the selected set of chemicals.

If the new *in vitro* models can be shown to predict human acute toxicity better than the HPV animal tests, *in vitro* batteries may totally replace the animal acute toxicity tests in further HPV testing.

11.0 MEIC Evaluation Guidelines Checklist

A complete and formal assessment of the validation status of MEIC in regard to the ICCVAM evaluation guidelines would require the following to be reviewed and evaluated:

ICCVAM Evaluation Guidelines

1.0 Introduction and Rationale of each Test Method
1.1 Scientific basis for each test method
1.1.1 Purpose of each proposed method, including the mechanistic basis
1.1.2 Similarities and differences of modes and mechanisms of action in each test system as compared to the species of interest (e.g., humans for human health-related toxicity testing).
1.2. Intended uses of each proposed test method.
1.2.1 Intended regulatory use(s) and rationale.
1.2.2 Substitute, replace, or complement existing test methods.
1.2.3 Fits into the overall strategy of hazard or safety assessment. If a component of a tiered assessment process, indicate the weight that will be applied relative to other measures.
1.2.4 Intended range of materials amenable to test and/or limits according to chemical class of physico-chemical factors.
2.0 Proposed Each Test Method Protocol(s)
2.1 Detailed protocol for each test method, duration of exposure, know limits of use, and nature of the response assessed, including:
2.1.1 Materials, equipment, and supplies needed
2.1.2 Suggested positive or negative controls.
2.1.3 Detailed procedures for conducting the test
2.1.4 Dose-selection procedures, including the need for any dose range-finding studies or acute toxicity data prior to conducting the test, if applicable;
2.1.5 Endpoint(s) measured
2.1.6 Duration of exposure
2.1.7 Known limits of use
2.1.8 Nature of the response assessed
2.1.9 Appropriate vehicle, positive and negative controls and the basis for their selection
2.1.10 Acceptable range of vehicle, positive and negative control responses
2.1.11 Nature of the data to be collected and the methods used for data collection
2.1.12 Type of media in which data are stored
2.1.13 Measures of variability
2.1.14 Statistical or non-statistical method(s) used to analyze the resulting data (including methods to analyze for a dose response relationship). The method(s) employed should

be justified and described
2.1.15 Decision criteria or the prediction model used to classify a test chemical (e.g., positive, negative, or equivocal), as appropriate
2.1.16 Information that will be included in the test report
2.2 Basis for each test system
2.3 Confidential information
2.4 Basis for the decision criteria established for each test
2.5 Basis for the number of replicate and repeat experiments; provide the rationale if studies are not replicated or repeated
Basis for any modifications to each proposed protocol that were made based on results from validation studies
3.0 Characterization of Materials Tested
3.1 Rationale for the chemicals/products selected for evaluation. Include information on suitability of chemicals selected for testing, indicating any chemicals that were found to be unsuitable
3.2 Rationale for the number of chemicals that were tested
3.3 The chemicals/products evaluated, including:
3.3.1. Chemical or product name; if a mixture, describe all components.
3.3.2 CAS number(s)
3.3.3 Chemical or product class
3.3.4 Physical/chemical characteristics
3.3.5 Stability of the test material in the test medium
3.3.6 Concentration tested.
3.3.7 Purity; presence and identity of contaminants.
3.3.8 Supplier/source of compound.
3.4 If mixtures were tested, constituents and relative concentrations should be provided whenever possible
3.5 Describe coding used (if any) during validation studies.
4.0 Reference Data Used for Performance Assessment
4.1 Clear description of the protocol for the reference test method. If a specific guideline has been followed, it should also be provided. Any deviation should be indicated, including the rationale for the deviation.
4.2. Provide reference data used to assess the performance of the proposed test method.
4.3 Availability of original datasheets for the reference data
4.4 Quality of the reference test data, including the extent of GLP compliance and any use of coded chemicals.
4.5 Availability and use of relevant toxicity information from the species of interest.
5.0 Test Method Data and Results
5.1 Complete, detailed protocol used to generate each set of data for each proposed test method.

- Any deviations should be indicated, including the rationale for the deviation. Any protocol modifications made during the development process and their impact should be clearly stated for each data set.
- 5.2 Provide all data obtained using each proposed test method. This should include copies of original data from individual animals and/or individual samples, as well as derived data. The laboratory's summary judgement as to the outcome of each test should be indicated. The submission should also include data (and explanations) from unsuccessful, as well as successful, experiments.
- 5.3 Statistical approach used to evaluate the data from each proposed test method
- 5.4 Provide a summary, in graphic or tabular form, of the results.
- 5.5 For each set of data, indicate whether coded chemicals were tested, experiments were conducted blind, and the extent to which experiments followed GLP procedures.
- 5.6 Indicate the lot-to-lot consistency of the test materials, the time frame of the various studies, and the laboratory in which the study or studies were done. A coded designation for each laboratory is acceptable.
- 5.7 Any data not submitted should be available for external audit, if requested

6.0 Test Method Performance Assessment

- 6.1 Describe performance characteristics (e.g., accuracy, sensitivity, specificity, positive and negative predictivity, and false positive and negative rates) of each proposed test method separately and in combination compared with the reference test method currently accepted by regulatory agencies for the endpoint of interest. Explain how discordant results from each proposed test were considered when calculating performance values.
- 6.2 Results that are discordant with results from the reference method.
- 6.3 Performance characteristics of each proposed test method compared to data or recognized toxicity from the species of interest (e.g., humans for human health-related toxicity testing), where such data or toxicity classification is available. In instances where the proposed test method was discordant from the reference test method, describe the frequency of correct predictions of each test method compared to recognized toxicity information from the species of interest.
- 6.4 Strengths and limitations of the method, including those applicable to specific chemical classes or physical/chemical properties
- 6.5 Salient issues of data interpretation, including why specific parameters were selected for inclusion

7.0 Test Method Reliability (Repeatability/Reproducibility)

- 7.1 Rationale for the chemicals selected to evaluate intra- and inter-laboratory reproducibility for each test method, and the extent to which they represent the range of possible test outcomes.
- 7.2 Analyses and conclusions reached regarding inter- and intra-laboratory repeatability and reproducibility for each test method
- 7.3 Summarize historical positive and negative control data for each test method, including number of trials, measures of central tendency and variability.

8.0 Test Method Data Quality

8.1 Extent of adherence to GLPs

- 8.2. Results of any data quality audits
- 8.3 Impact of deviations from GLPs or any non-compliance detected in data quality audits

9.0 Other Scientific Reports and Reviews

- 9.1 All data from other published or unpublished studies conducted using the proposed test method should be included.
- 9.2 Comment on and compare the conclusions published in independent peer-reviewed reports or other independent scientific reviews of the test method. The conclusions of such scientific reports and/or reviews should be compared to the conclusions reached in this submission. Any other ongoing evaluations of the method should be mentioned.

10.0 Animal Welfare Considerations (Refinement, Reduction, and Replacement)

10.1 Describe how the proposed test methods will refine (reduce pain or distress), reduce, and/or replace animal use compared to the current methods used.

11.0 Other Considerations

- 11.1 Aspects of test method transferability. Include an explanation of how this compares to the transferability of the reference test method.
 - 11.1.1 Facilities and major fixed equipment needed to conduct the test.
 - 11.1.2 Required level of training and expertise needed for personnel to conduct the test.
 - 11.1.3 General availability of other necessary equipment and supplies.
- 11.2 Cost involved in conducting each test. Discuss how this compares to the cost of the reference test method.
- 11.3 Indicate the amount of time needed to conduct each test and discuss how this compares with the reference test method.

12.0 Supporting Materials

- 12.1 Provide copies of all relevant publications, including those containing data from the proposed test method or the reference test method.
- 12.2 Include all available non-transformed original data for both each proposed test method and the reference test method.
- 12.3 Summarize and provide the results of any peer reviews conducted to date, and summarize any other ongoing or planned reviews.
- 12.4 Availability of laboratory notebooks or other records for an independent audit. Unpublished data should be supported by laboratory notebooks.

12.0 MEIC Related Publications (in chronological order)

Bernson, V., Bondesson, I., Ekwall, B., Stenberg, K., and Walum, E. (1987) A multicentre evaluation study of in vitro cytotoxicity. ATLA, 14, 144-145.

Bondesson, I., Ekwall, B., Stenberg, K., Romert, L. and Walum, E. (1988) Instruction for participants in the multicentre evaluation study of in vitro cytotoxicity (MEIC). ATLA, 15, 191-193.

Bondesson, I., Ekwall, B., Hellberg, S., Romert, L., Stenberg, K., and Walum, E. (1989) MEIC - A new international multicenter project to evaluate the relevance to human toxicity of in vitro cytotoxicity tests. Cell Biol. Toxicol., 5, 331-347.

Ekwall, B. (1989) Expected effects of the MEIC-study. In Report from The MEIC In Vitro Toxicology Meeting, Stockholm 9/3 1989, (Eds. T. Jansson and L.Romert), pp 6-8, Swedish National Board for Technical Development.

Ekwall, B., Gómez-Lechón, M.J., Hellberg, S., Bondsson, I., Castell, J.V., Jover, R., Högberg, J., Ponsoda, X., Stenberg, K., and Walum, E. (1990) Preliminary results from the Scandinavian multicentre evaluation of in vitro cytotoxicity (MEIC). Toxicol. In Vitro, 4, 688-691.

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Appendix I First Fifty Reference Chemicals

Acetaminophen

Aspirin

Ferrous sulfate

Diazepam

Amitriptyline

Digoxin

Ethylene glycol

Methyl alcohol

Ethyl alcohol

Isopropyl alcohol

1,1,1-Trichloroethane

Phenol

Sodium chloride

Sodium fluoride

Malathion

2,4-Dichlorophenoxyacetic acid

Xylene Nicotine

Potassium cyanide

Lithium sulfate Theophylline

Dextropropoxyphene HCl

Propranolol HCl

Phenobarbital

Paraquat

Arsenic trioxide

Cupric sulfate

Mercuric chloride

Thioridazine HCl

Thallium sulfate

Warfarin

Lindane

Chloroform

Carbon tetrachloride

Isoniazid

Dichloromethane

Barium nitrate

Hexachlorophene

Pentachlorophenol

Varapamil HCl

Chloroquine phosphate

Orphenadrine HCl

Quinidine sulfate

Diphenylhydantoin

Chloramphenicol

Sodium oxalate

Amphetamine sulfate

Caffeine

Atropine sulfate

Potassium chloride

Appendix II: Descriptions of the Essential Traits of 67 in vitro Methods

Method	hod							
° N	No. PIO	Cell type/ test system	Tissue of origin	Species	Endpoint	Incub- ation time	Testing laboratory ^b	Refer-
Hun	Human cell lines	lines						
-	Ξ	Hep G2	Hepatoma	Human	Protein content/Lowry	24 hours	Dierickx	မ
io	111:2	Hep G2	Hepatoma	Human	Protein content/ Sulphorhodamine B	24 hours	Hall, Cambridge & James	O
ω	11:2	Hep G2	Hepatoma	Human	MTT	24 hours	Gómez-Lechón, Jover,	3, 12
*	=	WI-1003/Hep G2d	Lung/Hepatoma	Human	Morphology	24 hours	Ponsoda & Castell' Garza-Ocañas & Torres-Alanis	۵
.55	II:3	Chang liver cells	Liver	Human	Morphology	24 hours	Garza-Ocañas & Torres-Alanis	ω
6	II:5	HeLa	Cervical carcinoma	Human	Morphology	24 hours	Ekwall & Malmsten	ω
.7	H:6	Hep 2	Epithelial carcinoma	Human	Protein content/	24 hours	Stammati, Zucco, Zanetti &	ω
900	II:7	Hep 2	Epithelial carcinoma	Human	LDH release	24 hours	Stammati, Zucco, Zanetti &	ω
۵	ė	HL-80	of larynx Promyelocutic	Human	ATP content	94 house	De Angelis	۵
,		***************************************	leukaemia	Tuman	VII. COURSE	8 annous	Sasaki & Ono	G
0.	III:10	HFL1	Fetal lung cells	Human	MTT	24 hours	Barile & Sookhoo*	5, 13
Ξ	III:11A SQ-5	SQ-5	Lung squamous	Human	LDH content	48 hours	Ohno, Wang, Sasaki & Hirano	3, 14
12.	III:12	SQ-5	Lung squamous	Human	Killing index ⁶	48 hours	Ohno, Wang, Sasaki & Hirano	3, 14
13.	II:10	NB-1	Neuroblastoma	Human	Protein content/	48 hours	Kunimoto, Miura, Aoki &	ω
∓	11:11	McCoy	Epithelial cells from	Human	Crystal violet staining Morphology/Trypan	72 hours	Kunimoto Shrivastava & Chevalier	ယ
5	13	an income cod	The state of the s					

Table I: Descriptions of the essential traits of 67 in vitro methods

Source: Clemedson et al. 1998. MEIC Evaluation of Acute Systemic Toxicity. Part IV. ATLA 26:131-183. (reprinted with permission from the editor)

112m	protocol								
Ø	XOLLIANI			Coomassie blue staining			cione of Hepa-1)		
	ట	Kärenlampi & Malmivuori	72 hours	Protein content/	Mouse	Hepatoma	Hepa-1c1c7 (Sub-	11:34	30.
	မ	Shrivastava & Chevalier	72 hours	Morphology/Trypan blue	Bovine	Kidney	MDBK	II:33	29
_	မ	Kunimoto, Miura, Aoki &	48 hours	Protein content	Rat	Pheochromocytoma	PC12h	11:32	28
_	۔ د	Romert, Jansson & Jenssen	48 hours		Mouse	Ascites sarcoma	BP8	11:31	27.
-	5	Hall, Cambridge & James	24 hours	Protein content/ Sulphorhodamine B	Pig	Kidney	LLC-PK1	III:40	26
3, 12		Gómez-Lechón, Jover, Ponsoda & Castell*	24 hours	MTT	Mouse	Fibroblasts	3T3	11:30	55
3, 16	ယ္ဆယ္	Ferro, Bassi & Canepa ^k Barile, Borges, Arjun &	24 hours 24 hours	Macromolecular content [3H]-proline uptake	Rat	Hepatoma Lung epithelial cells	HTC L2	II:23	23 24
_	3	Lewan & Andersson	10 minutes Lewan	ATP leakage	Mouse	Subline of Ehrlich	ELD	11:20	120
	ω	10 minutes Lewan & Andersson	10 minutes	ATP leakage	Mouse	Subline of Ehrlich	ELD	II:19	21.
							ll lines	Animal cell lines	Ani
	Verries 5	Valentino, Monaco, Pierngostini, Amati & Governa	3 hours	Estadum promide Locomotion stimulated by chemotactic peptide	Human	Blood	Polymorphonucleur leukocytes'	III:22	20.
0.	verna 5	Valentino, Monaco, Pieragostini, Amati & Governa	3 hours	Viable cell count fluorescein diacetate/	Human	Blood	III:21 Polymorphonuclear leukocytes'	III:21	19.
ı							Human primary cultures	nan pr	Hw
3, 15		Garza-Ocañas & Torres-Alanis Ekwall & Malmsten Dierickx	168 hours 168 hours 6 weeks	Morphology/pH changes pH changes (phenol red) Protein content/Lowry	Human Human Human	Liver Cervical carcinoma Epithelial cells from embryonic lung	Chang liver HeLa MRC-5 (finite cell line)	II:12 II:14 II:15	18.7.16

Table I: continued

Method	bod							
o.	No.	Cell type/ test system	Tissue of origin	Species	Endpoint	Incub- ation time	Testing laboratory ^b	Refer- ence
31.	II:35	3T3-L1 (Sub-	Embryonal	Swiss	Protein content/Kenacid	72 hours	Clothier	ω
20	38	clone of 3T3)	fibroblasts	mouse	blue staining			•
9	1	A31-1-1	whole emoryo	mouse	Colony formation	168 hours	Tanaka, Wakuri, Izumi, Sasaki & Ono	ω
Anin	al pri	Animal primary cultures						
33		Muscle cells	Skeletal muscle	Rat	Spontaneous contractility	1 hour	Gülden, Seibert & Voss	3, INVITTOX
	II:45A	Neurons	Embryonal forebrain	Chicken	Neutral red uptake	20 hours	Sawver & Weiss	protocol 93"
36.5	11:50	Hepatocytes"	Embryonal forebrain Liver	Chicken Male rat	MTT		Sawyer & Weiss Gómez-Lechón, Jover, Ponsoda & Castell*	3, 12
37.	11:51	Hepatocytes*	Liver	Male rat	Morphology/Trypan blue	24 hours	Shrivastava & Chevalier	ω .
38.	11:52	Erythrocytes	Peripheral blood	Balb/c	ATP content	24 hours	Tanaka, Wakuri, Izumi,	ω
39.		Muscle cells	Skeletal muscle	Rat	Intracellular creatine kinase activity	24 hours	Gülden, Seibert & Voss	3,
6.		Muscle cells	Skeletal muscle	Rat	Glucose consumption	24 hours	Gülden, Seibert & Voss	protocol 93" 3,
£		Muscle cells	Skeletal muscle	Rat	Spontaneous contractility 24 hours		Gilden Seihart & Voss	INVITTOX protocol 93**
								XOLLIANI

Appendix III: Oral LD50 Doses for Rat and Mouse and Mean Oral Lethal Doses for Humans and Toxicity Categories

Oral LD50 Doses for Rat and Mouse and Mean Oral Lethal Doses for Humans

Chemical	Chemical	Rat	LD50	Mous	se LD50	Ave. Hui	man Dose
Number		mg/kg	umol/kg	mg/kg	umol/kg	mg/kg	umol/kg
28	Mercuric chloride	1	4	6	22	25.7	94.7
31	Warfarin	2	5	3	10	107.1	347.4
18	Potassium cyanide	5	77	9	131	2.9	43.9
26	Arsenic trioxide	15	74	31	159	4.1	20.9
30	Thallium sulfate	16	32	24	47	14.0	27.7
39	Pentachlorophenol	27	101	28	105	28.6	107.3
6	Digoxin	28	36	18	23	0.1	0.17
17	Nicotine	50	308	3	21	0.7	4.4
13	Sodium fluoride	52	1238	57	1357	92.8	2210.9
47	Amphetamine sulfate	55	149	24	65	20.0	54.3
38	Hexachlorophene	56	138	67	165	214.3	526.6
32	Lindane	76	261	44	151	242.9	835.1
21	Propoxyphene HCL	84	223	255	678	24.6	65.4
25	Paraquat	100	537	120	644	40.0	214.7
40	Varapamil HCL	108	220	163	331	122.3	249.1
23	Penobarbital	162	697	137	590	111.4	479.7
48	Caffeine	192	989	127	654	135.7	698.8
2	Acetylsalicylic acid	200	1110	232	1287	385.7	2140.5
20	Theophylline	244	1354	235	1304	157.1	872.1
42	Orphenadrine HCL	255	834	100	327	50.0	163.4
43	Quinidine sulfate	258	610	286	676	79.2	187.4
14	Malathion	290	878	190	575	742.8	2248.4
11	Phenol	317	3369	270	2869	157.2	1670.0
3	Ferrous sulfate	319	2100	680	4477	392.1	2581.0
5	Amitriptyline	320	1154	140	505	37.1	133.8
4	Diazepam	352	1236	45	159	71.4	250.8
37	Barium nitrate	355	1358	266	1016	37.1	142.1
15	2,4-Dichlorophenoxy-acetic acid	375	1697	347	1570	385.8	1745.3
22	Propamolol HCL	466	1575	320	1082	71.5	241.7
27	Cupric sulfate	469	1880	502	2012	290.6	1163.6
19	Lithium sulfate	492	4478	1190	10,828	1065.5	9691.8
49	Altropine sulfate	585	864	456	674	1.7	2.5
41	Chloroquine phosphate	623	1208	500	969	84.3	163.4
33	Chloroform	908	7605	36	302	999.8	8375.2
29	Thioridazine HCL	995	2445	385	946	68.6	1684
35	Isoniazid	1250	9117	133	970	171.5	1250.4
36	Dichloromethane	1601	18,846	873	10,280	1386.2	16,321.7
44	Diphenylhydantoin	1635	6480	150	595	300.0	1189.1
34	Carbon tetrachloride	2350	15,280	8264	53,726	1314.4	8545.4
1	Paracetamol	2404	15,899	338	2235	271.4	1795.2
45	Chloramphenicol	2500	7735	1500	4641	285.7	884.0
50	Potassium chloride	2598	34,853	1499	20,107	285.5	3830.0
12	Sodium chloride	3002	51,370	4003	68,493	2287.3	39,138.9

Oral LD50 Doses for Rat and Mouse and Mean Oral Lethal Doses for Humans

16	Xylene	4299	40,490	2119	19,953	899.8	8474.6
7	Ethylene glycol	4698	75,684	5498	88,567	1570.9	25,304.8
8	Methanol	5619	175,327	7289	227,414	1569.0	48,954.2
9	Ethanol	7057	153,145	3448	74,837	4712.2	102,262.2
46	Sodium oxalate	11160	83,284	5095	38,019	357.1	2665.3
10	1,1,1-Trichloroethane	11196	83,927	7989	59,884	5707.6	42,785.8

Source: E. Walum. 1998. Acute oral toxicity. EHP 106:497-503. (reprinted with permission from the editor)

Oral LD50 Doses for Rat and Mouse and Mean Oral Lethal Doses for Humans

Chemical	Chemical		LD50	Mous	e LD50	Ave. Hui	man Dose
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28	Mercuric chloride	1	4	6	22	25.7	94.7
18	Potassium cyanide	5	77	9	131	2.9	43.9
6	Digoxin	28	36	18	23	0.1	0.2
30	Thallium sulfate	16	32	24	47	14.0	27.7
47	Amphetamine sulfate	55	149	24	65	20.0	54.3
39	Pentachlorophenol	27	101	28	105	28.6	107.3
26	Arsenic trioxide	15	74	31	159	4.1	20.9
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38	Hexachlorophene	56	138	67	165	214.3	526.6
42	Orphenadrine HCL	255	834	100	327	50.00	163.4
25	Paraquat	100	537	120	644	40.00	214.7
48	Caffeine	192	989	127	654	135.7	698.8
35	Isoniazid	1250	9117	133	970	171.5	1250.4
23	Penobarbital	162	697	137	590	111.4	479.7
5	Amitriptyline	320	1154	140	505	37.1	133.8
44	Diphenylhydantoin	1635	6480	150	595	300.0	1189.1
40	Varapamil HCL	108	220	163	331	122.3	249.1
14	Malathion	290	878	190	575	742.8	2248.4
2	Acetylsalicylic acid	200	1110	232	1287	385.7	2140.5
20	Theophylline	244	1354	235	1304	157.1	872.1
21	Propoxyphene HCL	84	223	255	678	24.6	65.4
37	Barium nitrate	355	1358	266	1016	37.1	142.1
11	Phenol	317	3369	270	2869	157.2	1670.0
43	Quinidine sulfate	258	610	286	676	79.2	187.4
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45	Chloramphenicol	2500	7735	1500	4641	285.7	884.0
16	Xylene	4299	40,490	2119	19,953	899.8	8474.6
9	Ethanol	7057	153,145	3448	74,837	4712.2	102,262.2
12	Sodium chloride	3002	51,370	4003	68,493	2287.3	39,138.9
46	Sodium chloride Sodium oxalate	11160	83,284	5095	38,019	357.1	2665.3
7		4698	75,684	5498		1570.9	25,304.8
	Ethylene glycol Methanol		,		88,567		
8 10		5619	175,327	7289	227,414 59,884	1569.0	48,954.2 42,785.8
	1,1,1-Trichloroethane	11196	83,927	7989	,	5707.6	
34	Carbon tetrachloride Walum. 1998. Acute oral toxic	2350	15,280	8264	53,726	1314.4	8545.4

Oral LD50 Doses for Rat and Mouse and Mean Oral Lethal Doses for Humans

Chemical	Chemical	Rat I	_D50	Mouse	e LD50	Ave. Hui	nan Dose
Number		mg/kg	umol/kg	mg/kg	umol/kg	mg/kg	umol/kg
6	Digoxin	28	36	18	23	0.1	0.2
17	Nicotine	50	308	3	21	0.7	4.4
49	Altropine sulfate	585	864	456	674	1.7	2.5
18	Potassium cyanide	5	77	9	131	2.9	43.9
26	Arsenic trioxide	15	74	31	159	4.1	20.9
30	Thallium sulfate	16	32	24	47	14.0	27.7
47	Amphetamine sulfate	55	149	24	65	20.0	54.3
21	Propoxyphene HCL	84	223	255	678	24.6	65.4
28	Mercuric chloride	1	4	6	22	25.7	94.7
39	Pentachlorophenol	27	101	28	105	28.6	107.3
5	Amitriptyline	320	1154	140	505	37.1	133.8
37	Barium nitrate	355	1358	266	1016	37.1	142.1
25	Paraquat	100	537	120	644	40.0	214.7
42	Orphenadrine HCL	255	834	100	327	50.0	163.4
29	Thioridazine HCL	995	2445	385	946	68.6	168.5
4	Diazepam	352	1236	45	159	71.4	250.8
22	Propamolol HCL	466	1575	320	1082	71.5	241.7
43	Quinidine sulfate	258	610	286	676	79.2	187.4
41	Chloroquine phosphate	623	1208	500	969	84.3	163.4
13	Sodium fluoride	52	1238	57	1357	92.8	2210.9
31	Warfarin	2	5	3	10	107.1	347.4
23	Penobarbital	162	697	137	590	111.4	479.7
40	Varapamil HCL	102	220	163	331	122.3	249.1
48	Caffeine	192	989	127	654	135.7	698.8
		244	1354		1304	157.1	872.1
20 11	Theophylline Phenol	317	3369	235 270	2869	157.1	1670.0
						171.5	1250.4
35	Isoniazid	1250	9117	133	970	214.3	526.6
38	Hexachlorophene	56 76	138 261	67	165	242.9	835.1
32	Lindane			44	151	242.9 271.4	835.1 1795.2
1 50	Paracetamol	2404	15,899	338	2235		
50	Potassium chloride	2598	34,853	1499	20,107	285.5	3830.0
45	Chloramphenicol	2500	7735	1500	4641	285.7	884.0
27	Cupric sulfate	469	1880	502	2012	290.6	1163.6
44	Diphenylhydantoin	1635	6480	150	595	300.0	1189.1
46	Sodium oxalate	11160	83,284	5095	38,019	357.1	2665.3
2	Acetylsalicylic acid	200	1110	232	1287	385.7	2140.5
15	2,4-Dichlorophenoxy-acetic	375	1697	347	1570	385.8	1745.3
3	Ferrous sulfate	319	2100	680	4477	392.1	2581.0
14	Malathion	290	878	190	575	742.8	2248.4
16	Xylene	4299	40,490	2119	19,953	899.8	8474.6
33	Chloroform	908	7605	36	302	999.8	8375.2
19	Lithium sulfate	492	4478	1190	10,828	1065.5	9691.8
34	Carbon tetrachloride	2350	15,280	8264	53,726	1314.4	8545.4
36	Dichloromethane	1601	18,846	873	10,280	1386.2	16,321.7
8	Methanol	5619	175,327	7289	227,414	1569.0	48,954.2
7	Ethylene glycol	4698	75,684	5498	88,567	1570.9	25,304.8
12	Sodium chloride	3002	51,370	4003	68,493	2287.3	39,138.9
9	Ethanol	7057	153,145	3448	74,837	4712.2	102,262.2
10	1,1,1-Trichloroethane	11196	83,927	7989	59,884	5707.6	42,785.8

Toxicity Categories

Category	Signal Word	Oral LD ₅₀ (mg/kg)	Dermal LD ₅₀ (mg/kg)	Inhalation LD ₅₀ (mg/L) ²	Oral Lethal Dose	Eye Irritation	Skin Irritation
I - Highly Toxic	DANGER, POISON (skull & crossbones), WARNING	0 to 50	0 to 200	0 to 0.05	A few drops to a teaspoonful	Corrosive (irreversible destruction of ocular tissue) or corneal involvement or irritation persisting for more than 21 days	Corrosive (tissue destruction into the dermis and/or scarring)
II - Moderately Toxic	CAUTION	>50 to 500	>200 to 2,000	> 0.05 to 0.5	Over a teaspoonful to one ounce	Corneal involvement or irritation clearing in 8-21 days	Severe irritation at 72 hours (severe erythema or edema)
III - Slightly Toxic	CAUTION	>500 to 5,000	>2,000 to 20,000	>0.5 to 2	Over one ounce to one pint	Corneal involvement or irritation clearing in 7 days or less	Moderate irritation at 72 hours (moderate erythema)
IV - Relatively Non-toxic	none	>5,000	>20,000	> 2	Over one pint to one pound	Moderate irritation at 72 hours (moderate erythema)	Mild or slight irritation at 72 hours (no irritation or slight erythema)

¹ EPA/OPP does not currently use the inhalation toxicity values in 40 CFR 150.10(h). Instead, OPP uses values that are from a 2/1/94 Health Effects Division paper entitled "Interim Policy for Particle Size and Limit Concentration Issues in Inhalation Toxicity Studies".

Sources:

- (1) U.S. EPA, Office of Pesticide Programs. Label Review Manual. Chapter 8: Precautionary Labeling. http://www.epa.gov/oppfead1/labeling/lrm/chap-0.8.htm.
- (2) National Ag Safety Database. Toxicity of Pesticides. http://www.cdc.gov/niosh/nasd/docs2/as18700.html.
- (3) 40 CFR 156.10(h) Labeling Requirements for Pesticides and Devices. Warnings and precautionary statements.

² Four hour exposure.

Appendix IV: Oral Acute Single Lethal Doses in Humans

						Ref	Dose value	Dose values (g)	g (g)					
ě	Chemical	MLD/	6	=	a	5	=	5	ä	17	=	5	Other	Mean
-	Paracetamol	ē	ē									1		1
ķo	Acetylsalicylic acid	₽₽	33.6	17.5	30 ¥	17.5	50 50 51	25 6	17.5			10		15 19
Ä	Fe ²⁺ in iron (II)	EĒ	5 35	17.5			8	; ;						22 23
ì	sulphate	MLD			2	5	50	10.7	11.5	7.7	23.2			=
÷	Diazepam	55									1			38 3
28	Amitriatelian	Ě	•					š				ю		• E5
	hydrochloride	MIP.	•		×2.1	ы	ы	-	1.75					9 PS
ġn.	Digoxin	ē				0.005		0.015	0.0075					3
7.	Ethylene glycol	5		Ξ	1001					Ξ	Ξ			0.0011
98	Methanol	5	Ξ	70	123		67	8	23 E	9	;	Ξ		===
φ	Ethanol	£	\$3.8 55.8	23.8 280	276			55	55		17.5	59		25
ō.	Isopropanol	5	32	90	Ē		9	188	5				98 (9, 59)	98
		MI.D					;	9		000		38		196
F	1,1,1-Trichloroethane	Đ										- 1		3
90	Phenol	Ş	3	12	>6.7								193 (60), 802 (61)	28
	chlorida	Ę.	5.5	5	ю	11.5					8.8	00		% =:
		5											210*	160

Table II: Oral acute single lethal doses in humans

Source: Ekwall et al. 1998. MEIC Evaluation of Acute Systemic Toxicity. Part V. ATLA 26:571-616. (reprinted with permission from the editor)

26. 27. 29. 30.	25 24 22 22 22	19. 19. 19. 20A. 20B.	5 4
Arsenic trioxide Copper (II) sulphate Mercury (II) chloride Thioridazine hydrochloride Thallium sulphate	Theophylline Dextropropoxyphene hydrochloride Propranolol hydrochloride Phenobarbital Paraquat	2,4-Dichloro- phenoxyacetic acid Xylene Nicotine Potassium cyanide Lithium Lithium sulphate	Sodium fluoride Malathion
£2£2£2£2£2	<u> </u>	egegegegege	£2£2
1 4.8 0.21	5 ∞ Ξ	28 ^d 5.6 120 0.060 0.050 0.25	7.5 60
0.23 2.1 0.5 0.85	2 ∞	0.045 0.20	1.6
0.12	0.5 >1 1.5	0.14 0.14	1.2
	0.75 E9.6	0.040	5
,	œ		80 CH
0.25 15 2.5 0.5	1.28 E5.1 7.5	24.1 19.4 0.060 0.005 0.20 9.4 ^r	7.5 1 17.5
0.33 0.2 15	0.78 ^h 4 7.5	0.045	7.5
1 1502	0.65 0.5	28 12.9 0.05 0.20	7.5
5	0.64 1.2	0.05	5
0.1 0.5	7 1.5 0.075	21.5 0.045 0.2	-
0.3* 3.5 (59)*	11 (63) 5.2 (9, 59)*	5.6 (9)*	70 (62) 25 (9, 59)*
0.29 0.18 14 9.3 1.5 0.5 4.2 0.98	11 5.4 0.71 0.86 5 7.8 4.8 2.5 0.18	277 539 531 0.036 0.201 0.21	25.29

Table II: continued

\$ \$ \$	39. 39.	35 32 32 32	Ž.
Chloroquine phosphate Orphenadrine hydrochloride Quinidine sulphate	Dichloromethane Barium nitrate Hexachlorophene Pentachlorophenol Verapamil hydrochloride	Warfarin Lindane Chloroform Carbon tetrachloride Isoniazid	Chemical
PEPEPE	#5#5#5#5#5#5	#efefefefefe	MLD/
2.8	ю	9 E E E E	5
1-3 00	ω N3	∞ 1	=
5.54	- 5 8 8	3.5 6.4	5
7.2 2.8	,	œ	Ref
			erence I
		32.84 8.75 8.44 8.75	Dose values (g) Reference numbers 3 14 15 16
ν _α ω ω ω	8.6 ⁴	8.4 7.5	16 ues (g)
% → .	3.9	7.5° 28 14.8	17
211.5	21 24	¥ %	5
	2.25	6.4 22	5
2.2 (9)*	2 (62)		Other
5.9 5.9 5.9	97 24 26 26 27 28 28 34	7.5 177 3.5 70 92 92 8.4	Mean

50. Potass		47. Amphetamine sulphate	46. Sodium		44. Diphen 45. Chlora
Potassium chloride	Atropine sulphate	ate	Sodium exalate		Diphenylhydantoin Chloramphenicol
P = E	E E		Ę	MILD	5 <u>8</u> 5
	E6.5	7.5			E7.5
	10	0.1	30		9.1
	E0.1	0.1	15		E21
	E0.2k	7.5			
		7.5			
	5	0.15			
E45	9	0.12	2		8
18*	900		23	:	20
16.2	01 10	3	15		
	10 0.075	0.25	51		100
24 (65)			5 (64)	10 (62), 26 (9)	
18	9.1 0.12 0.12	9.9	0 8 2 2 8 2 3 8 2	3 2	26.8 20

high variability as well as tolerance makes it difficult to establish human LD.

*POISINDEX", Information Systems (ed. B.H. Rumack & D.G. Spoerke), Micromedex (Denuer, CO, USA). "Extrapolated from animal dosage. "Geometric mean value, when the quotient between original values (range) is larger than ten.

*One death.

^{&#}x27;Two lethal poisonings.
'One survivor and one dead.

^{12.5}mg/kg lethal in 14 days (16).1g lethal in 13 days (17).
Several survivors.

 $LD = mean\ lethal\ dose;\ MLD = minimal\ lethal\ dose;\ E = extrapolated;\ nr = not\ reported.$ Very variable.

Appendix V: Clinically Measured Acute Lethal Serum Concentrations in Humans

							z	Co	Concentration References numbers	Concentrations (mg/l) ences numbers	mg/l)		Other	Mean con-
ò	Chemical	MIC 131	6	=	120	2	=	15	16	17	2	19	refer- ences	Mean con- centration (mg/ml)
-	Paracetamol	8	300*	300		300*						400		330
ķo	Salicylic acid	58	1300 ^b	160	300		300					600		950
မှ	Iron	58		1000	, 8	800	1000		:			90		3 a g
٠	Diazepam	55	; 5	3 5	o			o	ě.	3		20		20.6
Ş	Amitriptyline	M S S S	ě	on 8						,5 Sq. 18		10		N -1 . On On O
œ.	Digoxin	5	2						9	0.027	8	0.01		0.018
7.	Ethylene glycol	55	4370						9	4370	0.000	2000		3600
ġø.	Methanol	55	1750	1000				Š		1600	1800	800		1400
æ	Ethanol	58	5	5000	5		4000	5000		9	3	4500		1600
10.	Isopropanol	W C S	3400	2000	900			1500	2000		3/00	3000		2800 1800
12 =	1,1,1-Trichloro- ethane Phenol	M E M E	â						5				180 (26)*	96°
	The second secon						ŀ							

Table III: Clinically measured acute lethal serum concentrations in humans

Source: Ekwall et al. 1998. MEIC Evaluation of Acute Systemic Toxicity. Part V. ATLA 26:571-616. (reprinted with permission from the editor)

26. Arsenic 27. Copper	21. The 22. Dex ph 23. Pro 24. Phe 25. Par	16. 2,4-1 acc 17. Xylo 18. Nice 19. Cya 20. Lith	13. Sod 800 14. Flus 15. Mal
enic	Theophylline Dextropropoxy- phene Propranolol Phenobarbital Paraquat	2,4-Dichlorophenoxy acetic acid Xylene Nicotine Cyanide Lithium	Sodium in sodium chloride Fluoride Malathion
2 <u>8</u> 2	W C C C C C C C C C C C C C C C C C C C	NZSZSZSZSZ SZSZSZSZSZ	828282 WEST 200
	100° 6 16 115	2.5	10800
	183 3.3 80	24 10	10800
3.92	100		ω
		2	
	4. 100		10800
		6g. 3 3	
	135 64 3.3°	E50	14.2
	130° 1.8° 3° 117	3.1 ^d	14 ⁴ x
1.5	110		
os Ng	50 10 200	o	ω
		43 (66) 11 ^N	4.4 (26)*
2.5 2.5	150 79 8 1.9 6.4 3.9 136 100 2'	510 110 110 122 122 123 124	11000 ^r 8.6 nr E4.4 E0.35

Table III: continued

١														
							ဂ	ncentr	Concentrations (mg/l)	mg/l)				
		2					Re	fereno	References numbers	bers			Other	Mean con-
ò	Chemical	MLC	10	=	12	13	14	15	16	17	18	19	Ι.	(mg/ml)
28	Mercury	5	3							0.65*		12	14.3 (67) ^{4.8}	o io
29.	Thioridazine	58	0.22		20.1					7.1d		20		:
30.	Thallium	WI C	0.3							1.543				0.3
=	Warfarin	85 85									1074		110 (26) ⁴	E110*
33	Lindane	55							0.5	0.92*				080
ដ	Chloroform	55								1659		200		=
3	Carbon	55								2044		-		
35	Isoniazid	MIC		10						77*				_ 29
36	Dichloromethane	55										3000		- ×
37.	Barium	55								or or			97 (26) ^d	or 50
38	Hexachlorophene	55	35.6°							52				- -
39	Pentachlorophenol	MIC	ô							74*				40

, ê	Verapamil	MLC	34.					Ē	44				3.7
÷	Chloroquine	5	0,			œ		9	224		•		=
Ď	Orphenadrine	58	•	6					3.6*				÷ =
ŝ	Quinidine	58	o	16.8					14.6		6		స్త
4	Diphenylhydantoin	MIC	9 =		10		9		8		3		=
		S	8	50					3		90		55.0
ŝ	Chloramphenicol	WI2				75		68	æ 190				E190
6	Oxalate	55							20*		20	20 (26)*	20
7.	Amphetamine	5								•	ю		∞ ₽
\$	Caffeine	5	150			;				160d.	150		150
œ	Atropine	55				150		_	0.1344				E0 13
6	Potassium	5								352			370
			397						364				

E = estimated/extrapolated; $LC = mean\ lethal\ serum\ concentration$; $MLC = minimal\ lethal\ serum\ concentration$; $S/D = high\ survived\ and$ as judged from high survived concentrations in reference 16. *May include acute chronic dosage. *Peak concentration. **S/D: 90/170 = 130 mg/l (17). *Acute dosage. *In blood. *Represents acute on chronic dosage: no reports on single-dose lethal poisonings. *Plane 4 anaesthesia. *Value probably originating from forensic medicine data. *Reported value of 90mg/l, which seems too high. 'Grey baby syndrome.

Appendix VI: Post-Mortem Acute Lethal Concentrations in Humans

						Cor	Concentrations (mg/l)	ns (mg/l)			
		5				Re	Reference numbers	umbers		Other	Mean con-
, Z	Chemical	MLC	17	20	21	222	23	24	25	ences	(mg/ml)
-	Paracetamol	5	248		250	280*			160		230
iю	Salicylic acid	58	661	500	500	732	150	250	700		620
မ	Iron	25	9.0	35			450	450			22 52
٠	Diazepam	58	5	3		;				10 (68)	= =
50	Amitriptyline	MIC S	3.7	6.32	0.55g	5.56	1.5	1.75	ю	50 (69)	1.23
6	Digoxin	5	0.025	0.015	0.0103				0.015		0.016
7.	Ethylene glycol	58	2400	3000	2400		0.005	0.005			0.0038 2600
900	Methanol	58	1900	8	1900						1900
9	Ethanol	55	5500	9	4000				5000		4800 680
0	Isopropanol	ME	1500	3500	1000	4000					1500 1000
Ξ	1,1,1-Trichloroethane	55	126		180				316*		170
2	Phenol	5	49	90	10				90		76
3	Sodium in sodium chloride	MIC								13000 (26)*	13000

Source: Ekwall et al. 1998. MEIC Evaluation of Acute Systemic Toxicity. Part V. ATLA 26:571-616. (reprinted with permission from the editor)

5 5 5 5 7 5	2 4 2 4 5 1	9 9 8 7 6	5 =
Arsenic Copper Mercury Thioridazine Thallium	Theophylline Dextropropoxyphene Propranolol Phenobarbital Paraquat	2,4-Dichlorophenoxy- acetic acid Xylene Nicotine Cyanide Lithium sulphate	Fluoride Malathion
WEYNER SERVICE	WE WE WE WE	WP S S S S S S S S S S S S S S S S S S S	MIC S
4.0 3.3 4.0 5.1	150 4.7 14 97	464 43 29 24.7 31.9°	15 281
12.5° 12.5° 0.5	150 4.1* 10 80 35	10.9 16* 3.7	ю
25 5		13.4 ¹ 25 5	ယ
2.36* 0.58	7.7* 16 210	669 17.7* 7.6*	
5	86 -3 50		,
11.5	55 7 1.50	13.6	
0.6	7 125 2	10.9 3.7 35	ы
3.3 (27)° 2.4 (27)°			
0.00 A 7 A 7 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	150 7.9 11.8 1120 6	570 9.9 9.9 144	280 280

l'able IV: continued

l											
						Conc	Concentrations (mg/l)	s (mg/l)			
						Ref	Reference numbers	mbers		Other	Mean con-
o.	Chemical	MLC	17	20	22	22	23	24	25	ences	(mg/ml)
=	Warfarin	35			5		5	5	_	100 (28)	100
Ñ	Lindane	58	0 00+k		3				:		0 07
ಷ	Chloroform	52	64	390	īģ	29			390		38
Z	Carbon tetrachloride	56	274		260				150		230
ģš.	Isoniazid	MLC S	117		150"		100	100			130
<u>6</u>	Dichloromethane	25	364	280	395 ^b	496			280		360
37.	Barium	55	192							< 20***	9
38	Hexachlorophene	55	35	35					35		35
.39	Pentachiorophenol	55	107	6	4 %				6		46 6
ē	Verapamil	MIC S	=	6.4				2.5			2.5
=	Chloroquine	55	30.5	17.2*	≈ 10	11.2	5	ω	ω		37
2	Orphenadrine	55	20.6	6	4 9	16.7	7	<u>အ</u> ၈	6		4 12
3	Quinidine	55	5*	40	88	40	15		ô		22 12
#	Diphenylhydantoin	S S	54**	100	25		50	50	100		88
1											

Table V: Human kinetic data'

Appendix VII: Human Kinetic Data

ş Ξ 1,1,1-Trichloroethane Lithium sulphate 000 Complete Absorption in the gut Time to pend (ingestion) I hour? 0.5-> 4 hours* 7-24 hours 12-24 hours First-order 0.7, 6 and 53 hours 2.8 hours 58 hours** and 25 hour ₹ Free Free Free Free Restricted 77 Liver, kidney Liver, kidney Erythrocytes Refer 35 ä

Source: Ekwall et al. 1998. MEIC Evaluation of Acute Systemic Toxicity. Part V. ATLA 26:571-616. (reprinted with permission from the editor)

days 23 minutes and 5 Restricted? Liver ^{Ch}
nr Restricted Liver, kidney 0.35* Restricted Liver, kidney
0.6? Free None nr Restricted Muscle,
nri Pree Liver, kidney, (fat) 0.6 Free Liver, kidney, hung, skin
2.6 Free CNS, liver,
0.11* Restricted none nr Free CNS, liver,
4.6 Restricted Kidney, hea
18 Free CNS, lung.
> 1 Restricted Blood, kidne
2 Restricted Blood, liver
0.22 Restricted Liver, kidne
0.6 Free Liver Lung, liver,
4.3 Free CNS, liver,
16 Free CNS, liver,
0.5 Free None

Table V: continued

Š.	Chemical	Absorption in the gut	Time to peak (ingestion)	Kinetics	TW	돌	Passage of blood-brain barrier	Accumulation in vital organs	Blood protein binding	Refer-
÷	Chloroquine	Good	1-3 hours*	Triphasic	2, 7 and 45	2	Free	Heart, liver, kidney,	55-61%	16, 49
ŝ	Orphenadrine	Good	3 hours	• 4	15 hours	o	Free	CNS, liver, lung	20-95%	16, 50, 51
å	Quinidine sulphate	Good	> 2 hours*	First-order?	> 7.8 hours*	2	Restricted	Liver, kidney, hearth	£06-09	15, 16
*	Diphenyl- hydantein	Peor/good	30-120 hours*	Zero-order and first-	24-230 hours**	0.6	Free	Liver, kidney, CNS	**00	52
5	Chloramphenicul	Good	2-3 hours	First-order	2.5 hours	1.2	Free	Liver, kidney	55%	
56	Sodium oxnlate Amphetamine	Poor Complete	6 hours? 1-4 hours*	First-order? First-order?	4 hours?* 7-34 hours?	E0.4	E0.4* Restricted 3-6.1 Free	Kidney, liver Liver, kidney	16%	26, 64 15, 16
<u>\$</u>	Caffeine	Complete	1 hour	First-order?	9-16 hours*	0.6	Free	None (liver 2x)	35-60%	5
8	Potassium chloride	Complete	0.5 hours	Multiphasic	ne .	3	Free?	None	None	654
I										

"Data for the overdose situation are indicated by an asterisk". *Absorption: complete = 100% and rapid, good = 80%, moderate = 20-80%, and poor = 0-20%. 'One value indicates T'/2 of the elimination phase. Successive values represent separate phases (alpha, beta, etc.). 'Other than references 10, 11, 13, 14 and 17. 'Non-linear in overdose? 'Also a biotransforming organ. "POISINDEX", Information Systems (ed. B.H. Rumack & D.G. Spoerke), Micromedex (Denver, CO, USA). "Absorbed as acetylsalicylic acid. 'Due to corrosivity. 'Probably large, i.e. around 51/kg; "Early accumulation.' Documented first therapeutic doses, i.e. bioavailability is decreased by rapid binding in the liver of a large fraction of the absorbed dose (25-85%). For most such chemicals, passage of the intestinal mucosa is probably complete. However, the term between rapid and slow acetylators. "Alpha-phase: 3 hours in overdose. "Dose-dependent ability). "Slow accumulation. "Alpha phase: 2.9 hours. "Probably large Vd and protein binding. "pH-dependent. often used in this table, based on literature reports on the total absorption (the sum of intestinal passage and first pass reduction of bioavail-Information Systems (ed. B.H. Rumack & D.G. Spoerke), Micromedex (Denuer, CO, USA). Varies *Dependent on formulation

nr = non reported; CNS = central nervous system (brain); GIT = gastrointestinal tract (gut); T½ = plasma half-life; Vd = distribution vo

Appendix VIII: Peaks from Approximate 50% Lethal Concentration (LC50) Curves

MEIC evaluation part V: rodent and human toxicity data

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Table VI: Peaks from approximate 50% lethal concentration (LC50) curves*

						Case rep	orts	
No.		Time to peak (hours)	Peak conc. mg1	Type of curve	Sub- lethal	Lethal (clinical)	Lethal (post- mortem)	Total
1.	Paracetamol	4	358	LC50	81	62	0	143
2.	Salicylic acid	20	1070	LC50	31	46	1	78
3.	Iron	4	43.5	LC50	15	12	0	27
4.	Diazepam	2	19.9	LC100	4	0	0	4
5.	Amitriptyline	6	1.69	LC50	8	6	10	24
6.	Digoxin	3	0.071	LC50	15	9	1	25
7.	Ethylene glycol	2.5	1550	LC50	28	12	9	49
8.	Methanol	2	3790	LC50	76	37	7	120
9.	Ethanol	1	8440	LC50	20	1	143	164
10.	Isopropanol	1	4960	LC50	13	2	2	17
11.	1,1,1-Trichloro- ethane	1	231	LC50	3	0	2	5
12	Phenol	0.5	80	LC50	3	0	4	7
	Sodium in sodium chloride	5	11700	LC50	3	9	1	13
14.	Fluoride	3	19.4	LC0	3	3	7	13
15.	Malathion	5	1.88	LC0	2	1	11	14
16.	2,4-Dichloro- phenoxyacetic acid	14 d	1125	LC50	7	1	4	12
17.	Xylene	1	110	LC0	3	0	1	4
18.	Nicotine	0.5	13.5	LC0	1	1	3	5
19.	Cyanide	0.5	16.4	LC50	12	9	10	31
20.	Lithium	3	97.2	LC100	4 ⁶	0	0	4 ^b
21.	Theophylline	12	180	LC50	57	18	1	76
22.	Dextropropoxy- phene	2	8	LC0	2	1	6	9
	Propranolol	4	3.11	LC50	6	2	1	9
24.	Phenobarbital	15	230	LC50	20	1	0	21
25.	Paraquat	2.5	12.6	LC50	23	66	16	105
26.	Arsenic	4	1.65	LC50	10	8	3	21
27.	Copper	11	15.9	LC50	10	5	1	16
	Mercury	12	40.1	LC50	12	2	4	18
	Thioridazine	4	4.08	LC50	1	1	4	6
30.	Thallium	24	7.35	LC50	25	5	2	32

^{*}From reference 26.

Source: Ekwall et al. 1998. MEIC Evaluation of Acute Systemic Toxicity. Part V. ATLA 26:571-616. (reprinted with permission from the editor)

Documented single-dose cases (not overdose on previous medication).

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Table VI: continued

					Case rep	orts	
No. Chemical	Time to peak (hours)	Peak conc. mg/l	Type of curve	Sub- lethal	Lethal (clinical)	Lethal (post- mortem)	Total
31. Warfarin	6	200	LC0	3	0	0	3
32. Lindane	6	1.3	LC0	5 2	2	1	8
33. Chloroform	2	490	LC50	2	0	5	7
34. Carbon tetrachlor	ide 6	5.8	LC50	5	1	1	7
35. Isoniazid	3	167	LC50	24	3	4	31
36. Dichloromethane	3	344	LC0	0	0	9	9
37. Barium	2	305	LC100	9	Ö	Ö	9
38. Hexachlorophene	5	116	LC50	2	i	i	4
39. Pentachloropheno		79.1	LC50	ī	ō	3	4
40. Verapamil	2	13.2	LC50	10	9	4	23
41. Chloroquine	2	9.41	LC50	4	1	9	14
42. Orphenadrine	2	11.3	LC50	6	ī	8	15
43. Quinidine	6	26	LC50	4	2	ō	6
44. Diphenylhydantoi	n 34	202	LC50	13	ī	ŏ	14
45. Chloramphenicol	6	180	LC0	5	4	Ö	9
46. Oxalate	6	110	LC0	1	1	0	2
47. Amphetamine	2	15.5	LC50	î	î	5	7
48. Caffeine	3	179	LC50	6	ō	4	10
49. Atropine	3	4.05	LC100	2	ŏ	ō	2
50. Potassium	ĭ	375	LC0	4	3	ĭ	8
ov. I otassium	•	310	LOU	•	•	•	

^bDocumented single-dose cases (not overdose on previous medication).

a few organs are routinely screened for chemicals, such as blood, heart, liver, kidney, brain and lung. Thus, the information on body distribution is often limited to these organs.

The qualitative human toxicity data

The human toxicity data presented in Table IX are the result of a study of references 10–17, in a few instances supplemented by data from other sources. In the same way as the kinetic data in Table V, the toxicity data represent the sum of the information from all the handbooks consulted. The classification of lethal symptoms into main causes and other causes of death, as well as the classifi-

cation of lethal action into known, unknown and hypothetical mechanisms, represent judgements by the handbook authors. However, the lists of lethal symptoms in various handbooks have been extensively edited to provide uniform terminology. The handbook authors have used a plethora of terms for essentially the same type of event. To mention only one example, circulatory failure in Table IX stands for vascular collapse, vasomotor collapse, shock, circulatory shock, hypovolaemic shock, hypotensive shock, and so on.

Potentially the most controversial data in Table IX are those that are based on mecha-

Appendix IX: Human Acute, Single-Dose Toxicity Data

			Mean			Toxic	
S.	No. Chemical	Lethal symptoms*	time to death	Danger	Target organs	metab- olites*	Lethal mechanisms
-	Paracetamol	Hypoglyenemic cuma Liver failure M Kidney failure	3-5 days	3	Liver P Kidney P (CNS)	More toxic intracellular metabolites	Known: Covalent NAPQII lipid peroxidation
10	Acetylsalicylic acid	Metabolic acidosis M Cerebral bleedings Pulmonary oedema Cardiovascular failure	48 hours	2	Kidney P Liver P CNS P Lung P GIT P	Salicylic acid is the reactive metabolite of the parent compound	Known: General cell poison. Uncoupling of oxidative phosphorylation, inhibition of Kreb's cycle dehydrogenases
ω	Iron (II) sulphate	Haematemesis GIT perforation Pulmonary oedema CNS excitation/depression Circulatory failure Liver and kidney failure	6 or 48 hours	72 hours	GIT P Liver P Kidney CNS CVS Lung P	ŧ.	Known: General cell poison. Inhibition of oxidative phosporylation and ATP; lipid peroxidation
*	Diazepam	CNS depression M	2 hours	3 hours	CNS	(Nordiazepam) Unknown	Unk
çn	Amitriptyline hydrochloride	CNS excitation/ depression Heart arrythmias/arrest M	< 12 hours	6 days	CNS Heart	(Nortriptyline) Hypothetical: Blocks noradrenaline, 5-HT and dopamine presynaptic uptake; prevents reuptake o heart noradrenaline	Hypothetical: Blocks noradrenaline, 5-H' and dopamine presynaptic uptake; prevents reuptake heart noradrenaline
ارة	Digoxin	Heart arrythmias/ arrest M Hyperkalaemia	7 hours	20 hours	GIT CNS	(Metabolites)	Known: Impairing ion transport and increasing sarcoplasmic Ca by binding to Nn/K ATPase, increasing automaticity of cell

Table IX: Human acute, single-dose toxicity data

Source: Ekwall et al. 1998. MEIC Evaluation of Acute Systemic Toxicity. Part V. ATLA 26:571-616. (reprinted with permission from the editor)

5	15	F	10.	90	po po	.7
Sodium chloride	Phenol	1,1,1-Tri- chloroethane	Isopropanol	Ethanol	Methanol	Ethylene glycol
CNS excitation/depression M Cerebral bleedings Cardiovascular failure Pulmonary oedema Vasculitis	CNS excitation/depression M Heart arrest/pulmonary oedema Liver and kidney failure	CNS depression M Heart arrythmias Cardiovascular failure Pneumonia	CNS depression M Cardiovascular failure Pneumonia	CNS depression M Cardiovascular failure	CNS depression M Metabolic acidosis Cardiovascular failure	1-12 hours: CNS excitation/depression M 12:24 hours: heart failure 24-72 hours: kidney failure
20 hours	1 hour	3 hours	3 hours	6 hours.	32 hours ^d 173 hours ^f	17 hours
25 hours	24 hours	4 hours	48 hours	12 hours	. 3	72 hours
CNS P Lungs Kidney VS P	CNS Heart Liver Kidney	CNS P CVS Lung P	CVS CVS Lung P	CVS	CNS P' Pancreas P Liver P Kidney P Heart P	CNS Heart P Kidney P
ę	f	£	ę,	(Acetaldehyde)	Formaldehyde Formic acid	Glyoxalate Glycolate Oxalate
Known: Acute dehydration of brain cells caused by osmotic shift of water to the outside of the blood-brain harrier	Known: General protoplasmic poison that denaturates proteins	Unknown	Unknown	Hypothetical: Interference with cell membrane fluidity, pertubing proteins, such as ion channels. Depression of postsynaptic potentials in CNS	Hypothetical: Accumulation of formic acid leads to metabolic acidosis. Lactate inhibits mitochondrial respiration	Hypothetical: Metabolites inhibit mitochondria, leading to metabolic acidosis. Oxalate decreases S-Ca
	18, 34		80			

Table IX: continued

9	5.	17. 3		5	¥	No.
Potassium cyanide	Nicotine	Xylene	2,4-Dichloro- phenoxyncetic acid	Malathion	14. Sodium fluoride	No. Chemical
CNS excitation/depression M Metabolic acidosis Circulatory failure	CNS excitation/depression M Cardiovascular failure	CNS depression M Heart arrythmias/arrest Heart failure Pulmonary oedema	Hyperthermia/myotonia CNS excitation/depression Metabolic acidosis Heart failure Liver failure	Early: Cholinergic crisis/ respiratory failure M Later: Heart failure Heart arrythmias/arrest	Cardiovascular failure CNS excitation/depression	Lethal symptoms*
0.5-1 hour	minutes -1 hour	1-2 hours?	8-96 hours	0.5-6 hours	2-4 hours	Mean time to death
4 hours	4 hours	72 hours	48 hours	24 hours	20 hours	Danger
CNS P Heart VS	PNS	CNS P Heart Lung P Liver P	CNS P Liver P Kidney P Heart	CNS Muscles Heart P	Hearth CNS ^h Liver Kidney	Target organs
e	, d	¢	ę	Maloxon	£	Toxic metab- olites ^b
Known: General enzyme inhibition. High affinity for ferric ion. Inhibits cytochrome oxidase	Known: Cholinergic block causing polarisation of CNS and PNS synapses	Unknown: Heart failure caused by sensitisation of myocardium to endogenous catecholamines?	Hypothetical: Hypermetabolism due to uncoupling of oxidative phosphorylation. Direct toxin to striated muscle	Known: Inhibition of acetylcholine esterase resulting in acetlycholine accumulation in CNS and effector organs	Hypothetical: Protoplasmic poison interfering with many enzymes. May lower S-Ca and induce potassium efflux from cells	Lethal mechanisms
		-	. •	•		Refer- ences

25	2	23	13	12	20
Paraquat	Phenobarbital	Propranolol hydrochloride	Dextropropoxy- phene hydrochloride	Theophylline	Lithium sulphate
Early (24 hours): CNS excitation Pulmonary oedema Heart failure Kidney failure M Liver failure Later (48 hours-6 days): Pulmonary fibrosis M	CNS depression M Circulatory failure	CNS excitation/depression Cardiovascular failure Bronchospasm	CNS excitation/depression Heart arrythmias/arrest Cardiovascular failure	CNS excitation M Metabolic acidosis Heart arrythmias Electrolyte disturbances GIT bleedings	CNS depression Circulatory failure Kidney failure
3 hours- 4 weeks	5 hours- 7 days	0.5-2 hours	0.5-2 hours	1-5 days	1-7 days
=	10 days	4-20 hours CNS Hear VS	24 hours	2	7 days
Lung P Kidney P Heart P Liver P CNS P	CNS Heart	s CNS Heart VS	CNS	CNS Heart (GIT)	CNS Heart Kidney
ę	ą	tp?	(Norprop- oxyphene)	ą	æ
Hypothetical: Multisystem failure due to depletion of superoxide disputase, formation of free-radicals, and lipid peroxidation. Lung fibrosis due to accumulation of paraquat in this oxygen-rich organ	Hypothetical: GNS depression through inhibition of GABA synapses? Inhibits hepatic NADH cytochrome oxidoreductase	Unknown: Beta-adrenergic blockade?	Hypothetical: Binds to morphine receptors. Stabilises cell membranes. Norpropoxyphene is a primary cardioloxin	Unknown: Inhibits prostaglandins and cGMP metabolism. Adenosine receptor antagonist	Unknown: Partial substitution for normal cations of cells may disturb energy processes?

Table IX: continued

30. Thallium sulphate	29. Thioridazine hydrochlorid	28. Mercury (II) chloride	27. Copper (II) sulphate	26. Arsenic trioxide	No. Chemical
	1				
Gastroenteritis Cardiovascular failure M Respiratory failure Kidney failure Liver failure	CNS depression Heart arrythmias/arrest M	Gastroenteritis Circulatory failure Kudney failure	Liver failure Kidney failure Intravascular haemolysis Circulatory failure CNS excitation depression	Gastroenteritis Circulatory failure Heart failure Pulmonary oedema Intravascular haemolysis Kidney failure Liver failure	Lethal symptoms*
24 hours-3 weeks	2-10 hours	3 hours-14 days	3 hours-7 days	l hour-4 days	Mean time to death
4–5 weeks	2	14 days	4 days	4 days	Danger
Heart P VS Kidney P Liver P CNS P	CNS	Kidney P VS GIT P	Liver P Kidney VS	Kidney P Heard Lawer P VS P CNS P GIT P	Target
ą	(Mesoridazine?) Unknown	ą	ę	ā	Toxic metab- olites ^b
Hypothetical: Enzyme inhibition by binding to sulphydryl groups of mitochondrial membranes. Interference with oxidative	Unknown	Hypothetical: Changes membrane potentials and blocks enzyme reactions in cells by targeting the sulphydryl part of active sites of some enzymes	Hypothetical: Cupric copper is reduced to cuprous form by thiol groups in cell membranes. Superoxide is formed by reoxidation of cuprous copper, which induces lipid peroxidation	Known: Cellular poison. Multisystem failure due to uncoupling of exidative phosphorylation and inhibition of pyruvate and succinate exidative pathways	Mechanisms
š			. =	-	Refer- ences

(Carbon monoxide/
(Intracellular metabulites)
More toxic intracellular metabolites?
More toxic intracellular metabolites?
r _p ,
(Metabolites?)

Table IX: continued

2 9	No. Chemical	Lethal symptons	Mean time to death	Danger	Target organs	Toxic metab- olites ^b	Lethal
37.	Barium nitrate	Muscle paralysis/ respiratory failure Heart arrythmias/arrest High blood pressure Convulsions	2-3 hours or 2-3 days	24 hours	Muscle" Heart (Kidney)	ą	Hypothetical: Neuromuscular depolarisation. Potassium is forced into cells by an action on Na/K ATPase?
, s	Hexachlorophene Early: Gastro Hyper Circul 12–18 48–60 arryth	Early: Gastroenteritis Hyperthermin Circulatory failure 12–18 hours: CNS excitation/depression 48–60 hours: Heart arrythmias/arrest	4-60 hours	3 days	GIT VS Heart CNS*	ą.	Hypothetical: Uncoupling of oxidative Uncoupling of oxidative phosphorylation in cells. Binding to proteins in cytoplasma membrane and cell organelles
9.9	Pentachloro- phenol	Hyperthermia CNS excitation/depression Circulatory failure Myotonia Metabolic acidosis	4-24 hours	24 hours	Heart P VS CNS Liver P Kidney P	ę	Hypothetical: Uncoupling of oxidative phosphorylation. Protein binding, including selective enzyme inhibition (liver/kidney P450)
	Verapamil hydrochloride	Circulatory failure Heart arrythmias/arrest Metabolic acidosis CNS depression Hypoglycaemia	24 hours	36 hours	VS Heart	(Metabolites)	Known: Inhibition of transmembrane Ca flux in excitatory tissues. Also alpha-adrenergic blocking
≑	Chloroquine phosphate	Cardiovascular failure Cardiac arrythmias/arrest M CNS excitation/depression Hypokalaemia	1-24 hours	24 hours	Heart VS CNS	ę,	Hypothetical: Stabilisation of cell membranes leading to reduction of excitation and conduction in heart. Interference with mitochondria

47.	\$	j \$	#	ء ا	į ŝ
Amphetamine sulphate	Sodium oxalate	Chloramphenicol	Diphenyl- hydantoin	Quinidine sulphate	Orphenadrine hydrochloride
(Hyperthermia) Cardiac arrythmias/arrest CNS excitation/depresson M Metabolic acidosis	Initially (minutes): Gastroenteritis Circulatory failure Later (hours): CNS excitation/depression Heart arrythmiss/arrest Later (2 days): Kidney failure	Cardiovascular failure CNS excitation/depression Metabolic acidosis (Liver and kidney failure)	(Nystagmus/ataxia) CNS excitation/depression M Heart arrythmias/arrest*	Early: Heart failure Heart arrythmias/arrest M Later: CNS excitation/depression Kidney failure	CNS excitation/depression (max. 2-5 hours) M Heart arrythmias (max. 12-18 hours) Heart failure Liver failure
2-4 hours	3 hours	5 hours-2 days	30 hours-14 days	6 hours?	1-48 hours
7	٦	3	14 days	3	24 hours
CNS Pe Heart P Liver P Kidney	GIT CNS ^h Heart ^h Kidney	Heart VS CNS Liver Kidney	CNS (Cerebellum) Heart	Heart VS CNS Kidney	CNS Heart Liver P
ę	£	ş	ą	ę,	45
Hypothetical: Release of biogenic amines (dopamine, norepinephrine) from nerve terminal stores. Direct action as false transmitter	Hypothetical: Calcium-complexing action, depressing the level of ionized calcium in hody fluids. The direct action on GIT, VS and kidney cannot explained that way. Corrosivity is not caused by acidity.	Hypothetical: Birds to mitochondrial ribosomes and inhibits enzymes synthesis, for example, enzymes necessary for oxidative phosphorylation	Unknown: Binds to specific receptors in neuronal cell membranes. Inhibits voltage-dependent sodium channels	Unknown: Decreased electrolyte permeability of cell membranes leading to depression of heart excitability, conduction velocity and contractility.	Unknown

Table IX: continued

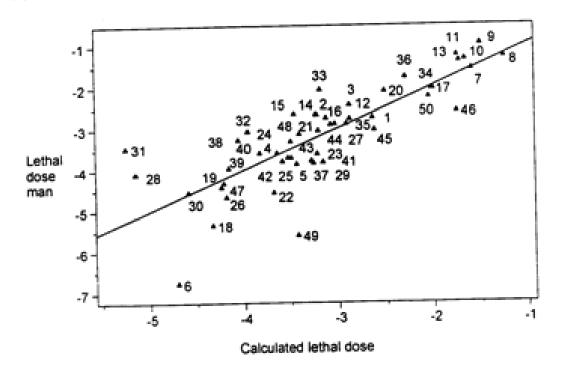
No. C	# 0		50. P
No. Chemical	Caffeine	Atropine	Potassium chloride
Lethal symptoms*	Initially (3 hours): Heart arrythmias/arrest Pulmonary oedema Later (3 hours-3 days): CNS excitation/depression	(Psychosis/hyperthermia) CNS excitation/depression Heart arrythmias/arrest M	CNS excitation/depression 2 hours Paralysis Heart arrythmias/arrest M
Mean time to death	3 hours-3 days	15 hours	2 hours
Danger	3	24-48 hours	2
Target organs	Heart CNS	CNS Heart PNS	Heart CNS (Muscle)
Toxic metab- olites ^b	ę.	ę	£
Lethal	Hypothetical: Inhibition of phosphodiesterase leading to AMP accumulation. Translocation of intracellular calcium? Adenosine receptor antagonism?	Known: Antimuscarinic, anticholinergic action. Competitive antagonism of acetylcholine at cardiac and CNS receptor sites	Known: Essential cellular electrolyte maintains normal trans- membrane potential, necessary for heart conduction
Refer- ences		19	

Arranged in order of appearance, when possible. Characteristic but non-lethal symptoms have generally been omitted. CNS excitation stands for seizures, and CNS depression stands for all phases of coma including final respiratory arrest. For chemicals with multisystem failure or a very rapid action, it is difficult to indicate the main cause of death. Metabolites with higher toxicity than the parent compound. Metabolites with the same toxicity as the parent compound are bracketed. TP indicates toxicity from the parent compound, only, 'Other than references 10-17. 'Post-mortem cases. 'Including the eye (blindness). 'Clinical cases. 'POISINDEX', Information Systems (ed. B.H. Rumack & D.G. Spoerke), Micromedex (Denver, CO, USA). 'Targets of a decreased blood calcium level? 'TOMES', Information Systems (ed. B.H. plates of muscles. Rumack & D.G. Spoerke), Micromedex (Denver, CO, USA). (Cerebral bleeding is most life-threatening, "Inhalation, "Ingestion, "Motor end "Repeated dermal exposure. "Intravenous administration. "Vasculitis, haemorrhages

 $M=main\ causes\ of\ death;\ P=histopathological\ organ\ lesions;\ CNS=central\ nervous\ system\ (brain\);\ CVS=cardiovascular\ system;\ tp=toxicity\ of\ VS=vascular\ system\ (blood\ vessels/capillaries);\ GIT=gastrointestinal\ tract\ (gut);\ PNS=peripheral\ nervous\ system;\ tp=toxicity\ of\ vessels/capillaries)$ parent compound only; nr = not reported

Appendix X: Plot of Acute Lethal Dosage in Humans Against Values Calculated by a PLS Model Based on Rat Oral LD50 and Mouse Oral LD50

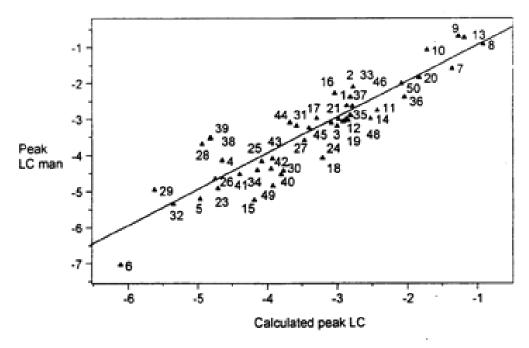
Figure 1: Plot of acute lethal dosage in humans against values calculated by a PLS model based on rat or al LD50 and mouse or al LD50.



Source: Ekwall et al. 1999. MEIC Evaluation of Acute Systemic Toxicity. Part VIII. (reprinted with permission from the editor)

Appendix XI: Plot of Peak Lethal Blood Concentrations in Man Against IC50 Values

Figure 10: Plot of peak tethal blood concentrations in man against IC-50 values calculated by a PLS model based on peak tethal blood concentrations in man, all 50 chemicals, and "blood-brain barrier compensated results" from assays 1, 5, 9 and 16.



Source: Ekwall et al. 1999. MEIC Evaluation of Acute Systemic Toxicity. Part VIII. (reprinted with permission from the editor)

Appendix XII: Priority Areas for Development and Evaluation of New In Vitro Tests

Table I: Priority areas for development and evaluation of new in vitro tests on systemic toxicity

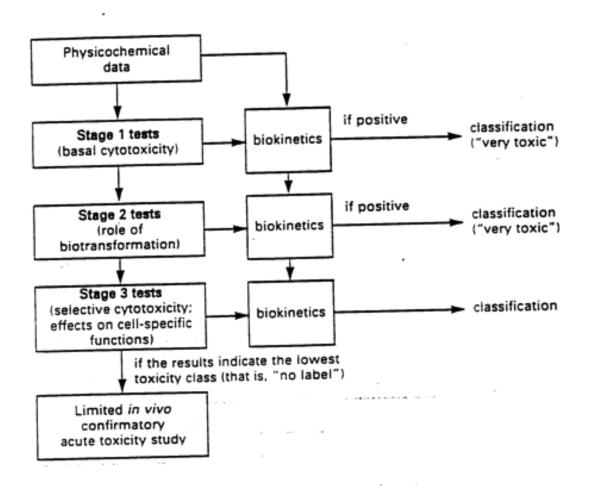
No. Subproject

- Repeat dose toxicity
- Mechanism studies:
 - a) protein denaturation
 - b) morphology of injury to cell lines
 - differential cytotoxicity 30 minutes/24 hours
 - d) toxicity to aerobic cells
 - e) time-frames for cytotoxic effects
- Extracellular receptor toxicity
- 4. Excitatory toxicity
- Reversibility of cytotoxicity
- 6. Passage across blood-brain barrier
- Absorption in the gut
- Blood protein binding
- Distribution volumes (Vd)
- More-toxic metabolites

Source: Ekwall et al. 1999. EDIT: A new international multicentre programme to develop and evaluate batteries of *in vitro* tests for acute chronic systemic toxicity. ATLA 27:339-349. (reprinted with permission from the editor)

Appendix XIII: Proposed Testing Scheme for the Classification and Labelling of Chemicals

Figure 1: Proposed testing scheme for the classification and labelling of chemicals according to their potential acute toxicities



Source: Ekwall et al. 1999. EDIT: A new international multicentre programme to develop and evaluate batteries of *in vitro* tests for acute chronic systemic toxicity. ATLA 27:339-349. (reprinted with permission from the editor)